

ASYMMETRIC SYNTHESIS OF α -ALKYLATED α -AMINO ACIDS: AZEPANE-2-CARBOXYLIC ACIDS

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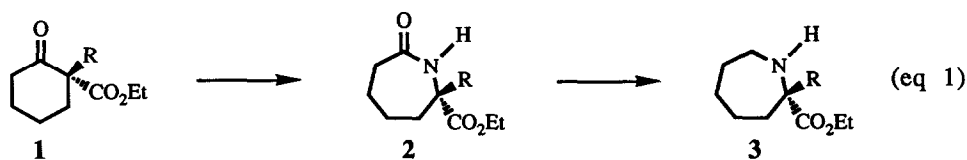
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Abstract: The synthesis of optically active α -alkylated azepane-2-carboxylic acid esters **3** was achieved via Schmidt rearrangement of optically active ethyl 2-oxo-1-alkyl-cyclohexanecarboxylates **1** followed by selective reduction of the amide carbonyl group.

The asymmetric synthesis of unusual and non-proteinogenic α -amino acids is of continuing interest because of their demonstrated or potential biological activity. α -Substituted α -amino acids belong to this group of compounds. They possess enzyme inhibitory properties² and are also of interest in peptide chemistry³ because the incorporation of α -alkylated α -amino acids into peptides restricts the available range of backbone conformations. Several elegant methods for the asymmetric synthesis of acyclic α -alkylated α -amino acids,⁴ carbocyclic α -amino acids,^{4,5} and α -substituted prolines^{4,6} have been developed. The development of asymmetric methodology for the synthesis of higher ring homologues of proline, however, has not yet received much attention. A recent report by Schöllkopf⁷ details the synthesis of α -alkylated pipecolic acids via the alkylation of mono-substituted bislactim ethers with dihalides, followed by an intramolecular ring closure.

We now wish to report on the first asymmetric synthesis of seven membered homologues of α -substituted prolines, the azepane-2-carboxylic acids. The methodology relies on ring expansion chemistry via the Schmidt rearrangement of optically active cyclic β -keto esters. We have previously reported⁸ on related chemistry for the synthesis of acyclic α -alkylated amino acids and are now extending this methodology toward the asymmetric synthesis of α -substituted azepane-2-carboxylic acids **3** (eq 1).



The synthesis starts with optically active ethyl 2-oxo-1-alkyl-cyclohexanecarboxylates **1**, which were obtained through diastereoselective alkylation⁹ of the lithio enamine of ethyl 2-oxo-cyclohexanecarboxylate utilizing the readily available *tert*-butyl ester of L-valine as

chiral auxiliary (Scheme 1, Table 1). The enantiomeric purity of the resulting β -keto esters was determined by $^1\text{H-NMR}$ experiments¹⁰ and found to be >95% ee for benzylic derivatives (entries 1-4, Table 1). Alkylation with methyl bromoacetate (entry 5, Table 1), however, gave only an enantiomeric excess of 59%.⁹

Scheme 1

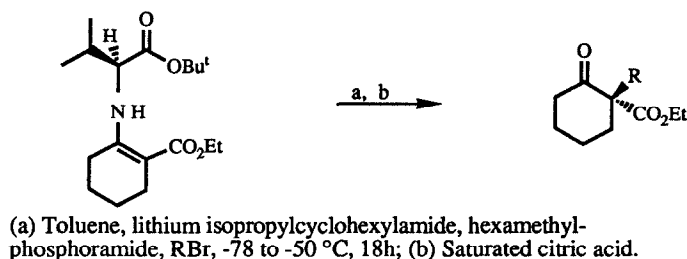


Table 1

entry	R	yield (%)	$[\alpha]_D^a$	ee % ^b
1	PhCH ₂ -	83	-110°	>95
2	(2-naphthyl)CH ₂ -	83	-116°	>95
3	<i>p</i> -BrPhCH ₂ -	72	-94°	>95
4	<i>m</i> -ClPhCH ₂ -	82	-60°	>95
5	CH ₃ O ₂ CCH ₂ -	68	-61°	59

^aThe optical rotations were taken in chloroform, $c \approx 1$. ^bDetermined by $^1\text{H-NMR}$ after the addition of tris[3-(heptafluoropropylhydroxymethylene)-(+)-camphorato] europium (III) derivative.

Schmidt rearrangement of ethyl 2-oxo-1-alkyl-cyclohexanecarboxylates **1** in chloroform with sodium azide (2-5 equiv) and methanesulfonic acid (9-11 equiv) afforded 7-ethoxycarbonyl-7-alkylazacycloheptan-2-ones **2** with retention of configuration⁸ and in excellent yield (Scheme 2, Table 2).

Scheme 2

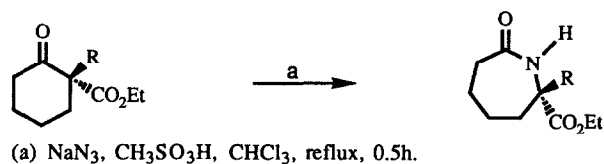


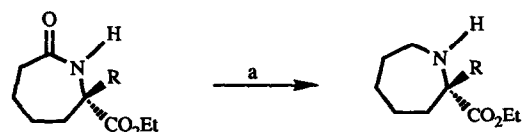
Table 2

entry	R	yield (%)	$[\alpha]_D^a$	e e % ^b
1	PhCH ₂ -	91	+4.6°	>95
2	(2-naphthyl)CH ₂ -	71	+23.9°	>95
3	<i>p</i> -BrPhCH ₂ -	81	-2.3°	not determined
4	<i>m</i> -ClPhCH ₂ -	85	-19.1°	>95
5	CH ₃ O ₂ CCH ₂ -	95	-3.8°	62

^aThe optical rotations were taken in chloroform, $c \cong 1$. ^bDetermined by ¹H-NMR after the addition of tris[3-(heptafluoropropylhydroxymethylene)-(+)-camphorato] europium (III) derivative.

Subsequent selective reduction¹¹ of the amide carbonyl group in the presence of an ester group with borane-methyl sulfide complex (3-3.5 equiv) in tetrahydrofuran completes the synthesis of the desired α -alkylated azepane-2-carboxylic acid ethyl esters¹² (Scheme 3, Table 3).

Scheme 3



(a) $\text{BH}_3 \cdot \text{CH}_3\text{SCH}_3$ (3-3.5 equiv), THF, -15 °C, 1 day.

Table 3

entry	R	yield (%)	$[\alpha]_D^a$	e e % ^b
1	PhCH ₂ -	56	-7.6°	>95
2	(2-naphthyl)CH ₂ -	64	-10.2° ^c	>95
3	<i>p</i> -BrPhCH ₂ -	67	-3.0°	>95
4	<i>m</i> -ClPhCH ₂ -	61	-10.5°	>95
5	CH ₃ O ₂ CCH ₂ -	62	-8.7°	59

^aThe optical rotations were taken in chloroform, $c \cong 1$. ^bDetermined by ¹H-NMR after the addition of tris[3-(heptafluoropropylhydroxymethylene)-(+)-camphorato] europium (III) derivative. ^c365 nm.

The extension of this methodology toward the asymmetric synthesis of α -alkylated pipecolic acids and higher homologues of azepane-2-carboxylic acids is under investigation.

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REFERENCES AND NOTES:

1. Current address: Chemical Process Development, Pharmaceutical Research and Development Division, Bristol-Myers Squibb Company, Syracuse, N.Y. 13221-4475
2. For reviews see: Silverman, R. B. *J. Enzyme Inhibition*, **1988**, *2*, 73. Walsh, C. T. *Ann. Rev. Biochem.* **1984**, *53*, 493.
3. For reviews see: Hruby, V. J.; Mosberg, H. I. In *Hormone Antagonists*; Agarwal, M. K., Ed.; Walter de Gruyter: Berlin, 1982; p 433. Venkataram Prasad, B. V.; Balaram, P. *CRC Crit. Rev. Biochem.* **1984**, *16*, 307. Weinstein, B. *Chemistry and Biochemistry of Amino Acids, Peptides and Proteins*; Marcel Dekker: New York, 1983; Vol. VII, pp 267-357. Tetrahedron Symposia-in-Print, Number 31; Hruby, V. J.; Schwyzler, R., Eds.; *Tetrahedron* **1988**, *44*, 661.
4. For recent reviews on the asymmetric synthesis of α -amino acids and α -substituted α -amino acids see: Williams, R. M. *Synthesis of Optically Active α -Amino Acids*; Pergamon: Oxford, 1989. Barrett, G. C. In *Chemistry and Biochemistry of the Amino Acids*; Barrett, G. C., Ed.; Chapman and Hall: London, 1985; p 246. Tetrahedron Symposia-in-Print, Number 33; O'Donnel, M. J. Ed.; *Tetrahedron* **1988**, *44*, 5253.
5. Vettiger, T.; Seebach, D. *Liebigs Ann. Chem.* **1990**, 195. Seebach, D.; Dziadulewicz, E.; Behrendt, L.; Cantoreggi, S.; Fitzi, R. *Liebigs Ann. Chem.* **1989**, 1215. Subramanian, P. K.; Woodard, R. W. *J. Org. Chem.* **1987**, *52*, 15. Schöllkopf, U.; Hauptreif, M. *Angew. Chem. Int. Ed. Engl.* **1986**, *25*, 192. Schöllkopf, U.; Hupfeld, B.; Gull, R. *Angew. Chem. Int. Ed. Engl.* **1986**, *25*, 754.
6. Moss, W. O.; Bradbury, R. H.; Hales, N. J.; Gallagher, T. *J. Chem. Soc., Chem. Commun.* **1990**, 51. Seebach, D.; Imwinkelried, R.; Weber, T. In *Modern Synthetic Methods*; Scheffold, R., Ed.; Springer: Berlin, 1986; pp 125-259.
7. Schöllkopf, U.; Hinrichs, R.; Lonsky, R. *Angew. Chem. Int. Ed. Engl.* **1987**, *26*, 143.
8. Georg, G. I.; Guan, X.; Kant, J. *Tetrahedron Lett.* **1988**, *29*, 403. For a related approach via the Curtius rearrangement see: Ihara, M.; Takahashi, M.; Niitsuma, H.; Taniguchi, N.; Yasui, K.; Fukumoto, K. *J. Org. Chem.* **1989**, *54*, 5413.
9. Tomioka, K.; Ando, K.; Takemasa, Y.; Koga, K. *J. Am. Chem. Soc.* **1984**, *106*, 2718. *Idem.* *Tetrahedron Lett.* **1984**, *25*, 5677. Tomioka, K.; Ando, K.; Yasuda, K.; Koga, K. *Tetrahedron Lett.* **1986**, *27*, 715. The absolute stereochemistry of the reaction products, obtained in the diastereoselective alkylation, was determined by conversion to compounds with known absolute stereochemistry.
10. The ^1H -NMR experiments were performed at 48 °C probe temperature. The absolute stereochemistry of the products is inferred by analogy to examples in reference 8 and 9 of this letter.
11. Brown, H. C.; Choi, Y. M.; Narasimhan, S. *J. Org. Chem.* **1982**, *47*, 3153.
12. All newly synthesized compounds were characterized by elemental analysis or HRMS and exhibited spectroscopic data in agreement with their structures.